



Novel Quantitative Method for Biodiesel Analysis



Georgia Institute of Technology

**North Avenue Trade School
opened in 1888 with 84 students**

**Over 17,000 students are
currently enrolled**

Sits on 400 acre campus

**Ranked # 8 among public
universities in the nation**

**Ranked in the top 5 for most
engineering disciplines**



Outline

Fossil fuels vs. Biofuels

Current biodiesel analytical methods

Developing a new analytical tool

Chemical approach

Phosphitylating agent

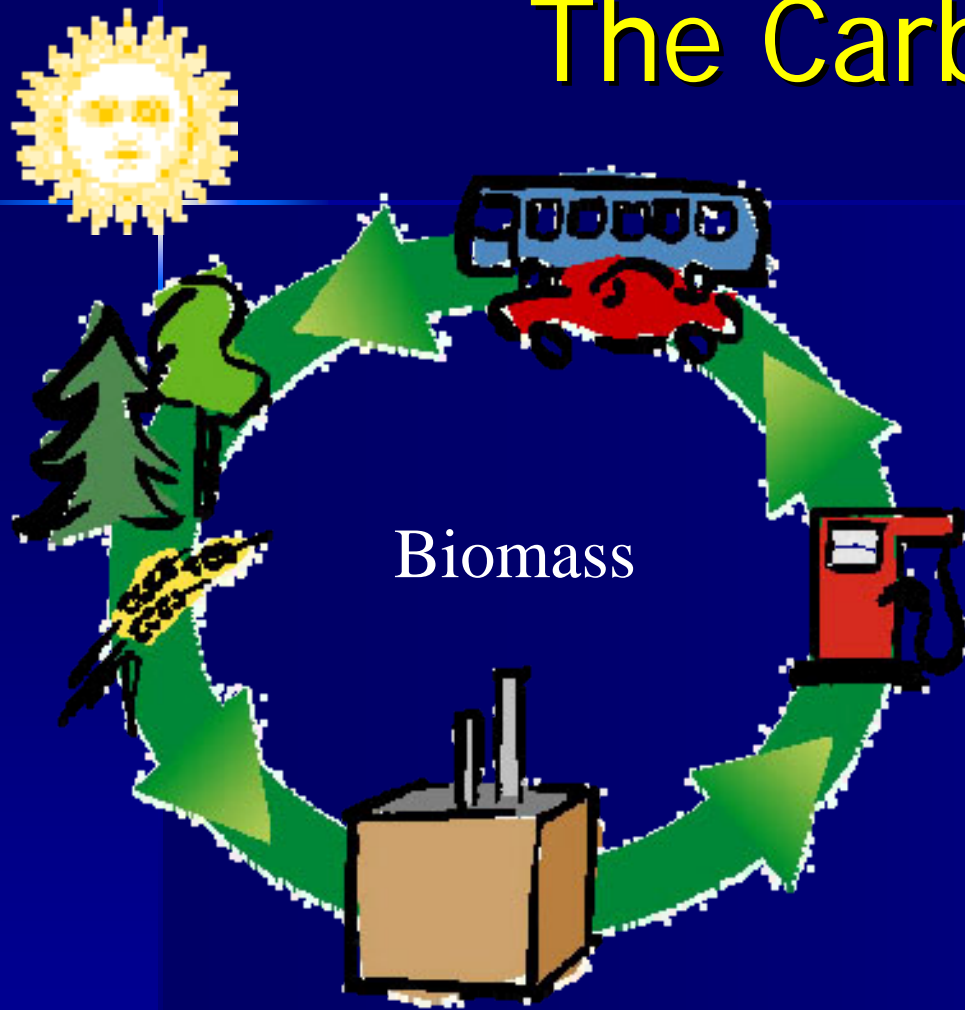
^{31}P -NMR chemical shift library

Process sample analysis

Method optimization

Rapid sample analysis

The Carbon-cycle

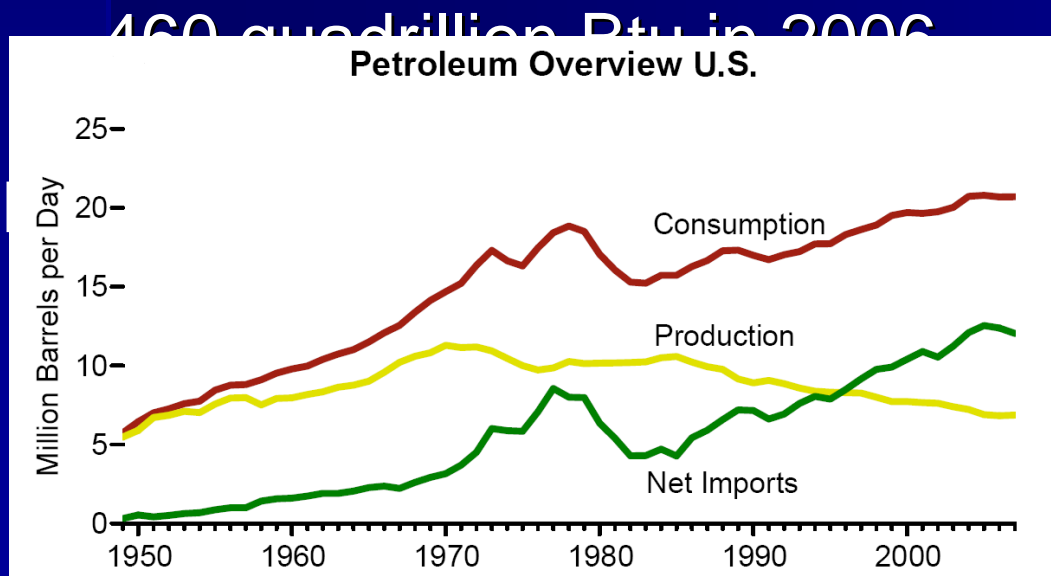
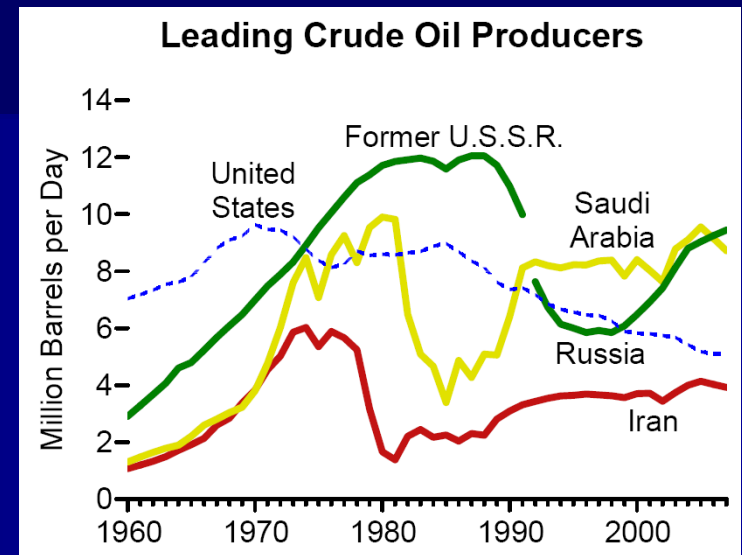
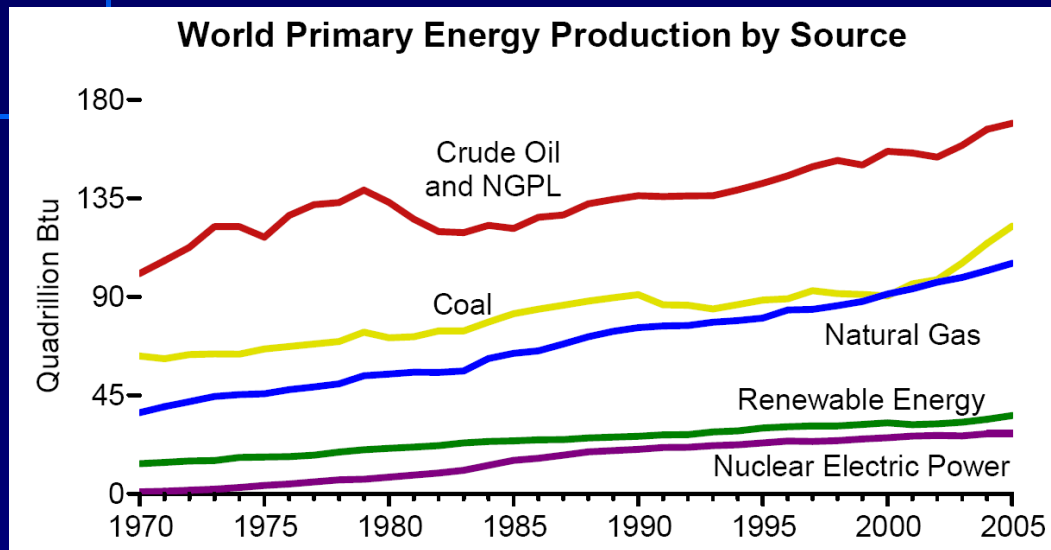


Closed cycle



Broken cycle

International Energy Overview



World crude oil production totaled 74 million barrels per day in 2008.

Need for Renewable Biofuels:

Energy security (Local)

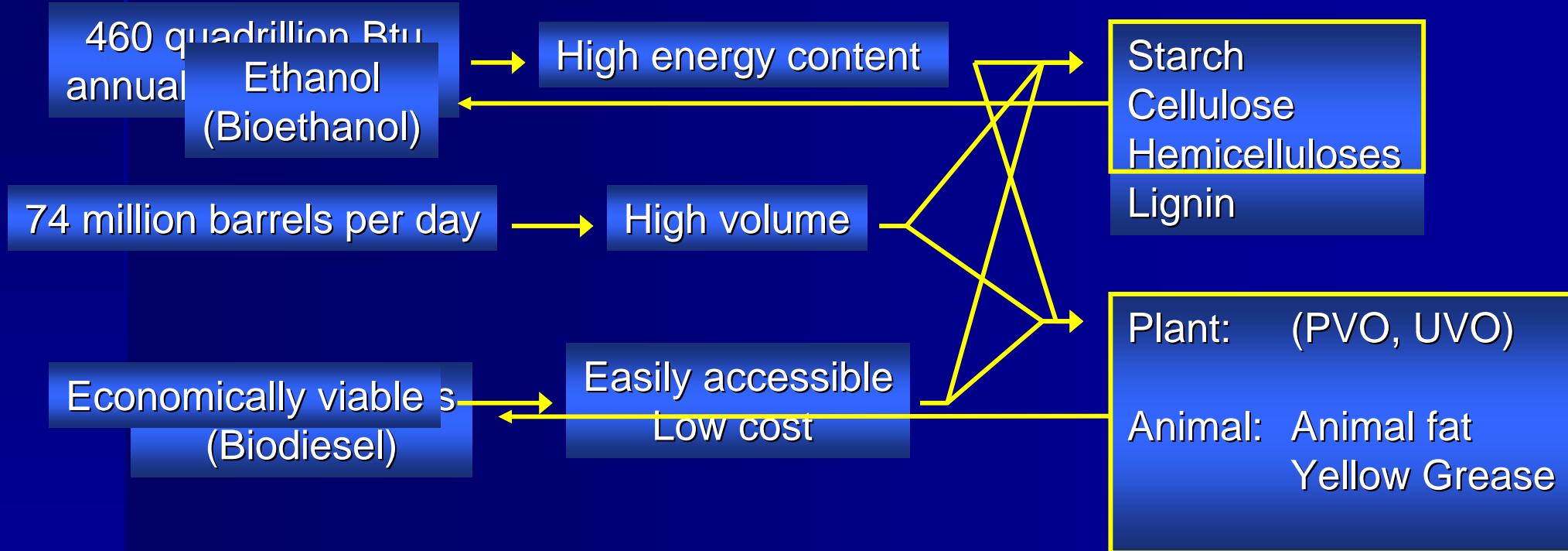
Carbon neutral/climate change

Sustainability

Potential Biofuel Precursors

Produced biofuels

Bio-renewable resources



Potential Biofuel Precursors

Produced biofuels

Bio-renewable resources

Ethanol
(Bioethanol)

Starch
Cellulose
Hemicelluloses
Lignin

Alkali fatty esters
(Biodiesel)

Plant: (PVO, UVO)
Animal: Animal fat
Yellow Grease

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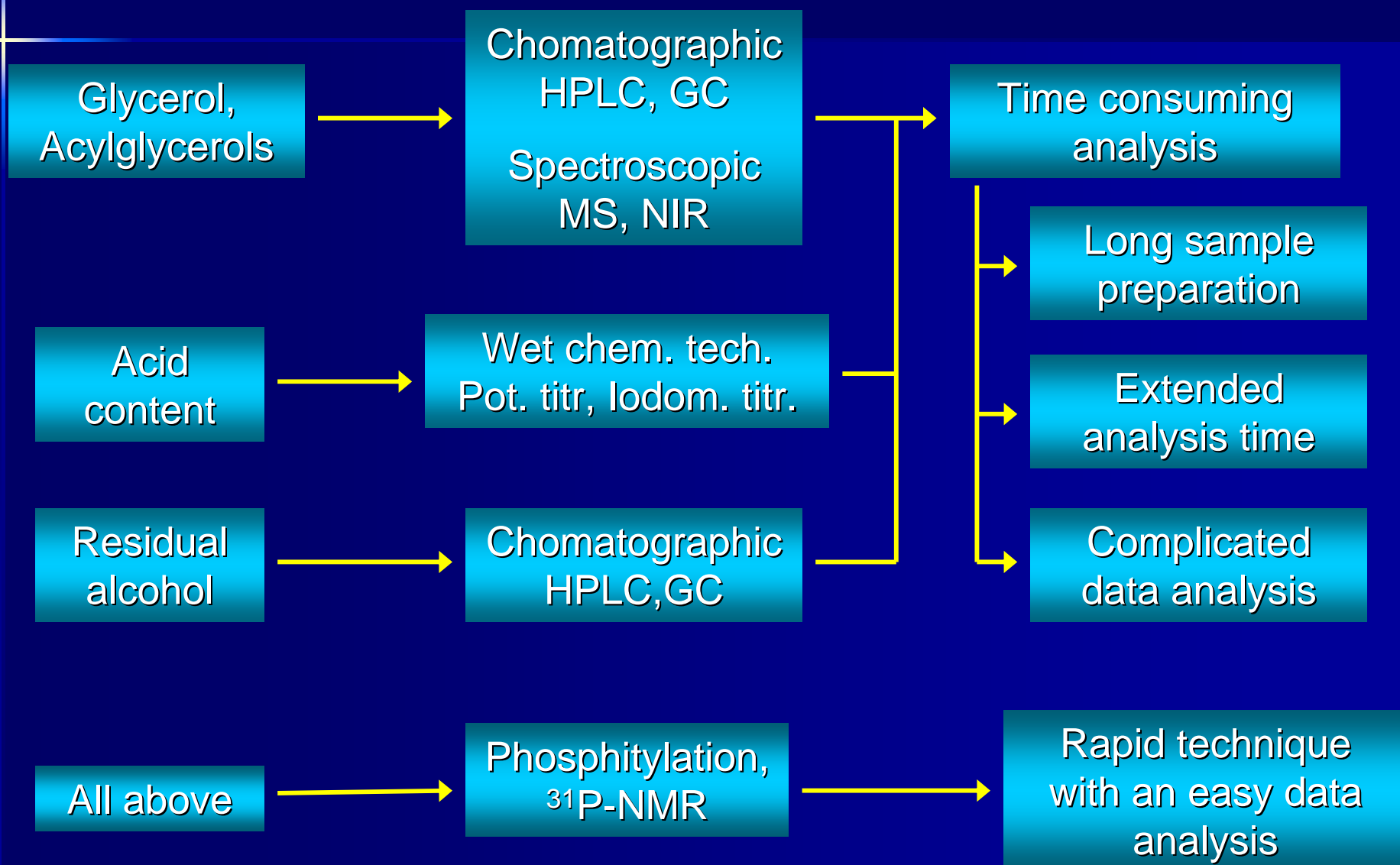
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Current Biodiesel Analytical Methods



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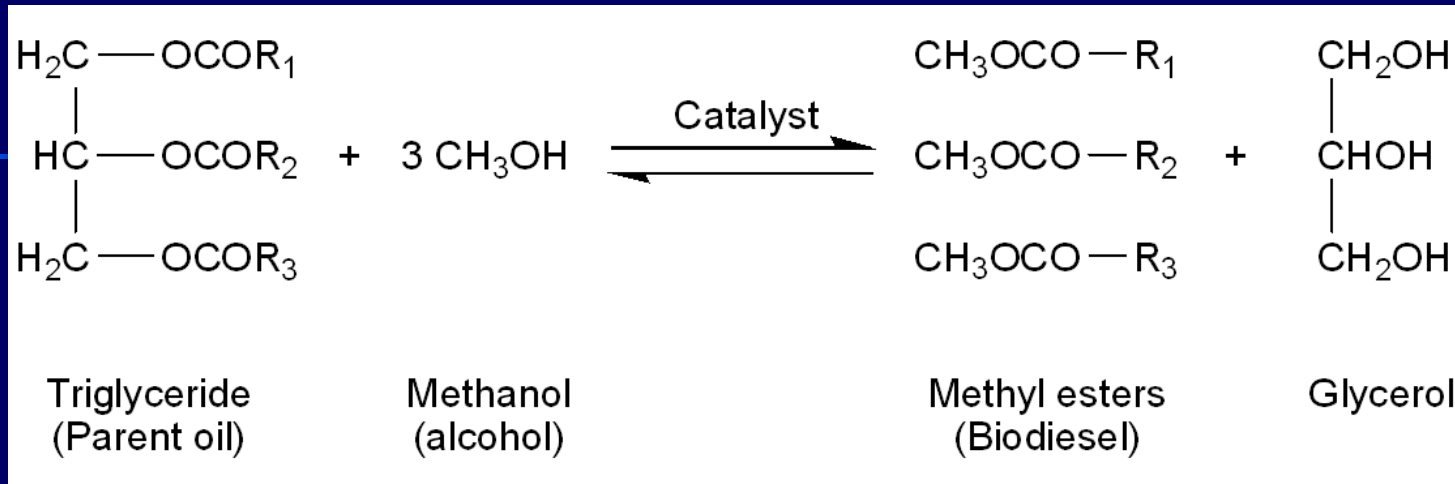
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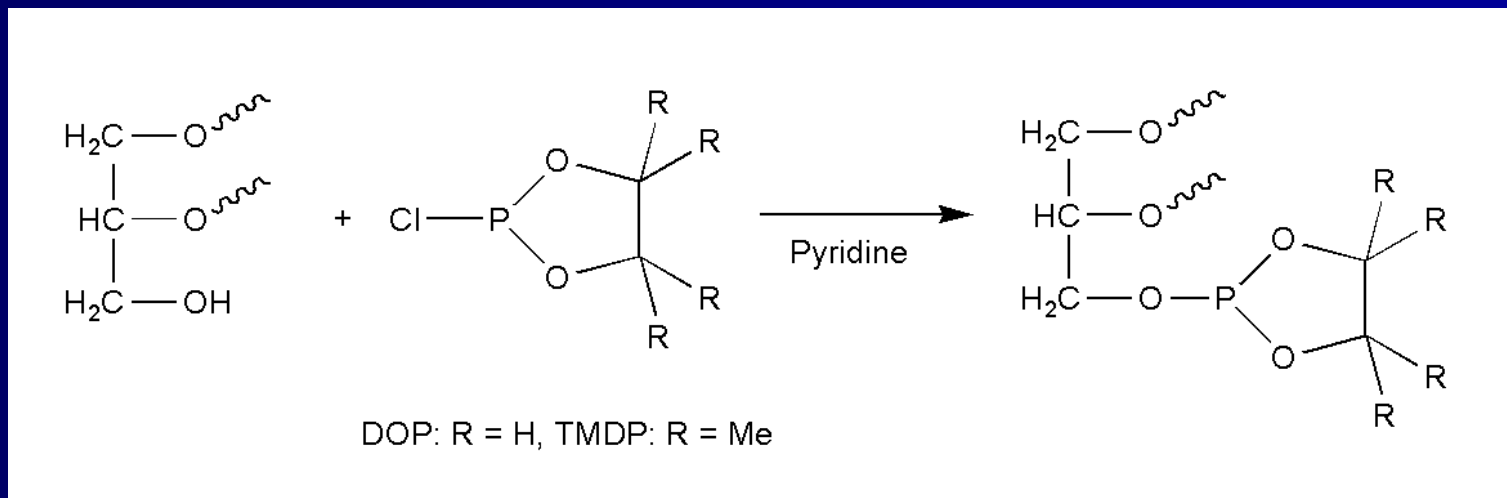
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Chemical Approach



Transesterification reaction for biodiesel production,
 R_{1-3} are hydrocarbon groups



Phosphitylation of partially substituted glycerols with DOP or with TMDP¹¹

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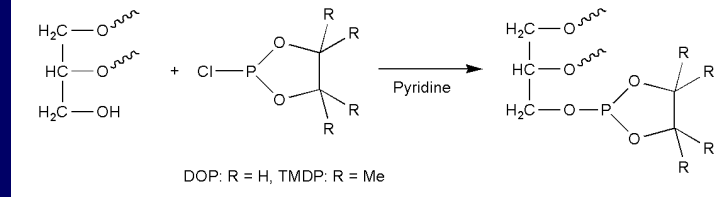
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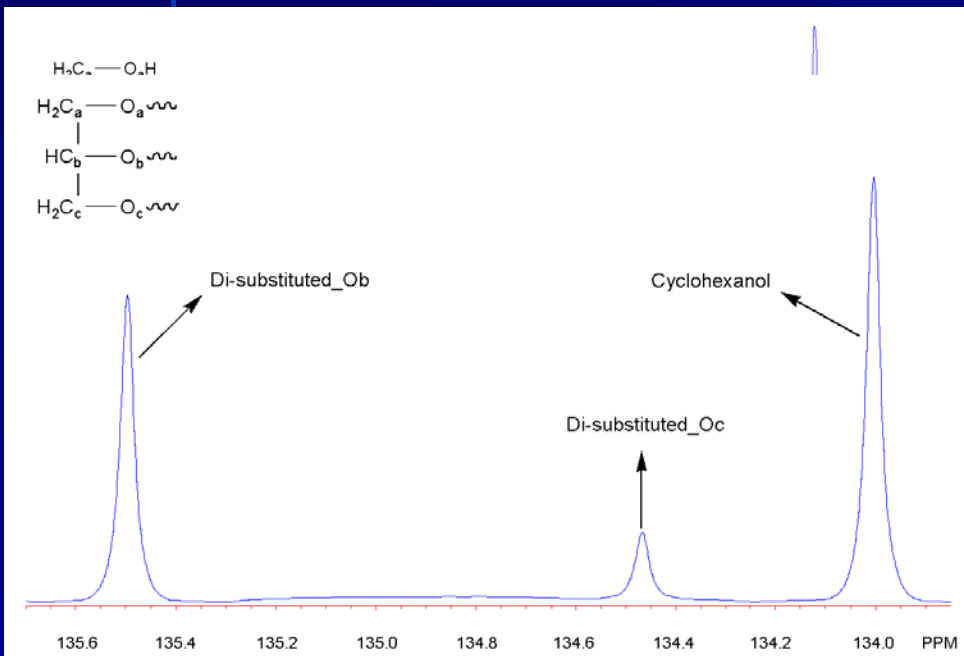
Phosphitylating Agent

DOP vs. TMDP

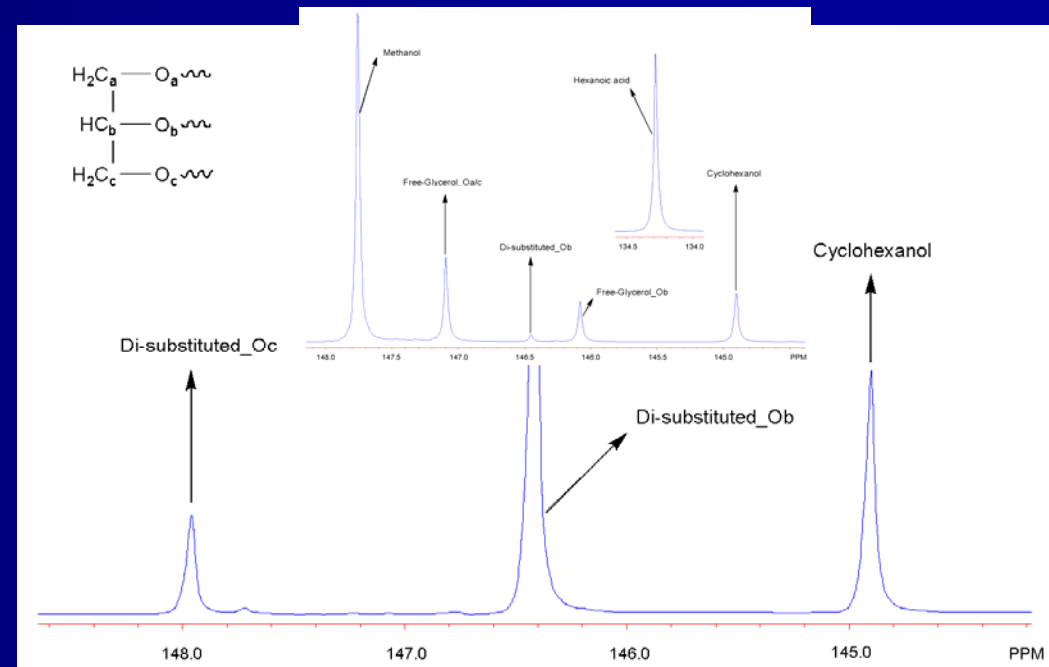


~85% 1,3-Dioleoylglycerol and ~15% 1,2-Dioleoylglycerol mixture

DOP



TMDP



TMDP provides greater spectral resolution of the individual hydroxyl groups of glycerol-derivatives → reagent of interest for additional studies.

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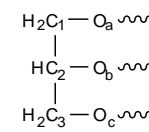
Rapid sample analysis

Biodiesel Analysis

³¹P-NMR chemical shift library

	³¹ P Signal (ppm)	Phosphityl. position
<i>Biodiesel production by-products</i>		
Free glycerol	147.1	O-a/c
	146.1	O-b
Methanol	147.8	
Ethanol	146.3	
Isopropanol	146.4	
<i>Analysis by-products</i>		
TMDP	174.7	
TMDP hydrolyzate	132.0	
Cyclohexanol (internal standard)	144.9	

$\begin{array}{c} \text{H}_2\text{C}_1-\text{OH} \\ \\ \text{HC}_2-\text{OH} \\ \\ \text{H}_2\text{C}_3-\text{OH} \end{array}$	Glycerol derivatives					Fatty acid
	Mono-substituted		Di-substituted		Tri *	
	Substituted position	1	1,2	1,3	1,2,3	
Phosphitylated hydroxyl	O-b	O-c	O-c	O-b	N/A	
Lipid chain						
<i>Lauric</i>	146.2	147.5	---	146.5	No signal	134.5
<i>Myristic</i>	146.2	147.5	---	146.4	No signal	134.5
<i>Palmitic</i>	146.2	147.6	147.9	146.4	No signal	134.4
<i>Stearic</i>	146.2	147.4	147.9	146.4	No signal	134.4
<i>Oleic</i>	146.1	147.4	147.7	146.4	No signal	134.4
<i>Linoleic</i>	146.2	147.4	147.7	146.4	No signal	134.3
<i>Linolenic</i> R—OH	---	---	---	---	---	134.3
³¹P Signal (ppm)	146.1-146.2	147.4-147.6	147.7-147.9	146.4-146.5	No signal	134.3-134.5



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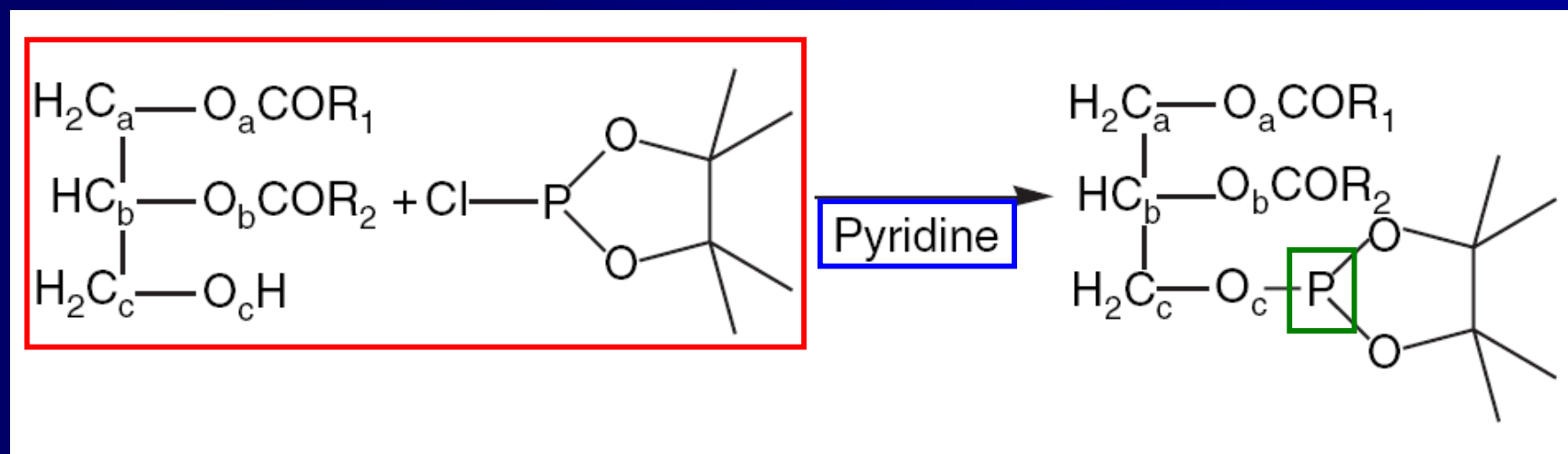
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TMDP/³¹P-NMR Method Optimization

Solvent system for biodiesel constituents



Phosphitylation of 1,2-diacylglycerol with 2-chloro-4,4,5,5-tetramethyl-1,3,2-dioxaphospholane at c-position. R₁₋₂ are hydrocarbon groups
Solvent mixture

The process sample and the reagent mixture has to be in solution

TMDP/³¹P-NMR Method Optimization

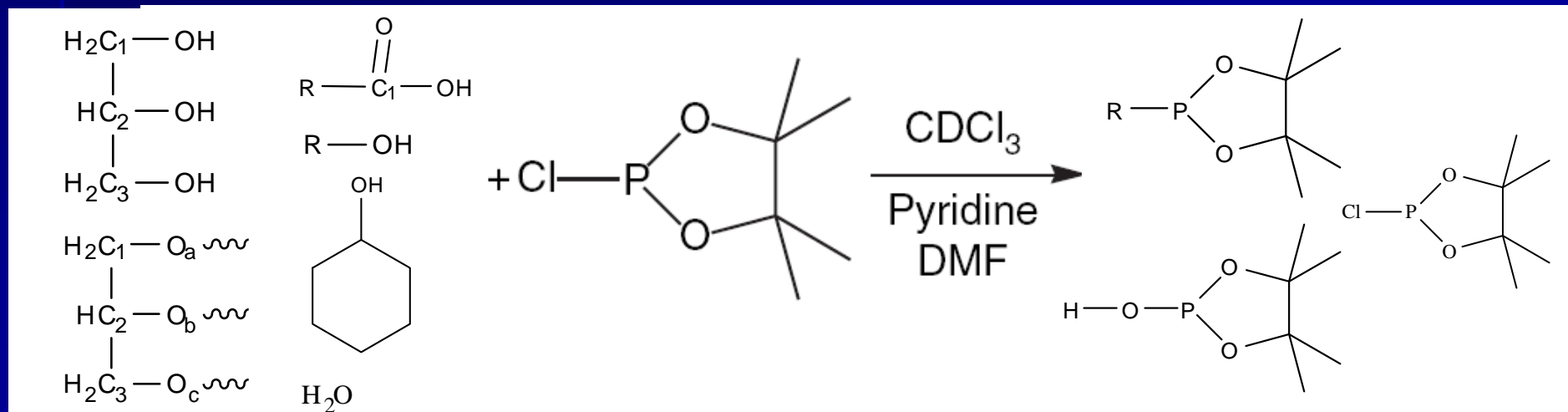
Spin-lattice relaxation time of biodiesel precursors

Rapid analytical method → Reduce analysis time to a minimum

³¹P spin-lattice relaxation time measurement

Anhydrous pyridine/CDCl₃/DMF in 1:1.2:1 ratio

Chromium acetylacetonate (Cr(acac)₃) as a relaxation agent (~3.60 mg/mL)
400 MHz, 25 °C, magnetic field of 9.4 T,



TMDP/³¹P-NMR Method Optimization

Optimized TMDP/³¹P-NMR analysis protocol

Solvent mixture

Anhydrous pyridine/CDCl₃/DMF in 1:1.2:1 ratio
~3.60 mg/mL of Cr(acac)₃ (*relaxation agent*)
~4.00 mg/mL of cyclohexanol (*internal standard*)

Optimized NMR pulse program (400 MHz, at 25 °C)

5 sec pulse delay, inverse-gated decoupling (Waltz-16), 90° pulse angle, a time domain of 327 K with one degree of zero filling, 4.0 Hz line broadening, and 16 acquisition transients
Phosphitylation
Acquisition process sample (150 μL) or Glycerol process sample (10 μL)

Solvent mixture 500 μL → Mix
TMDP (100 μL)

Sample preparation → Under 5 min

Analysis time → 56 sec

Data analysis → ~30 minutes

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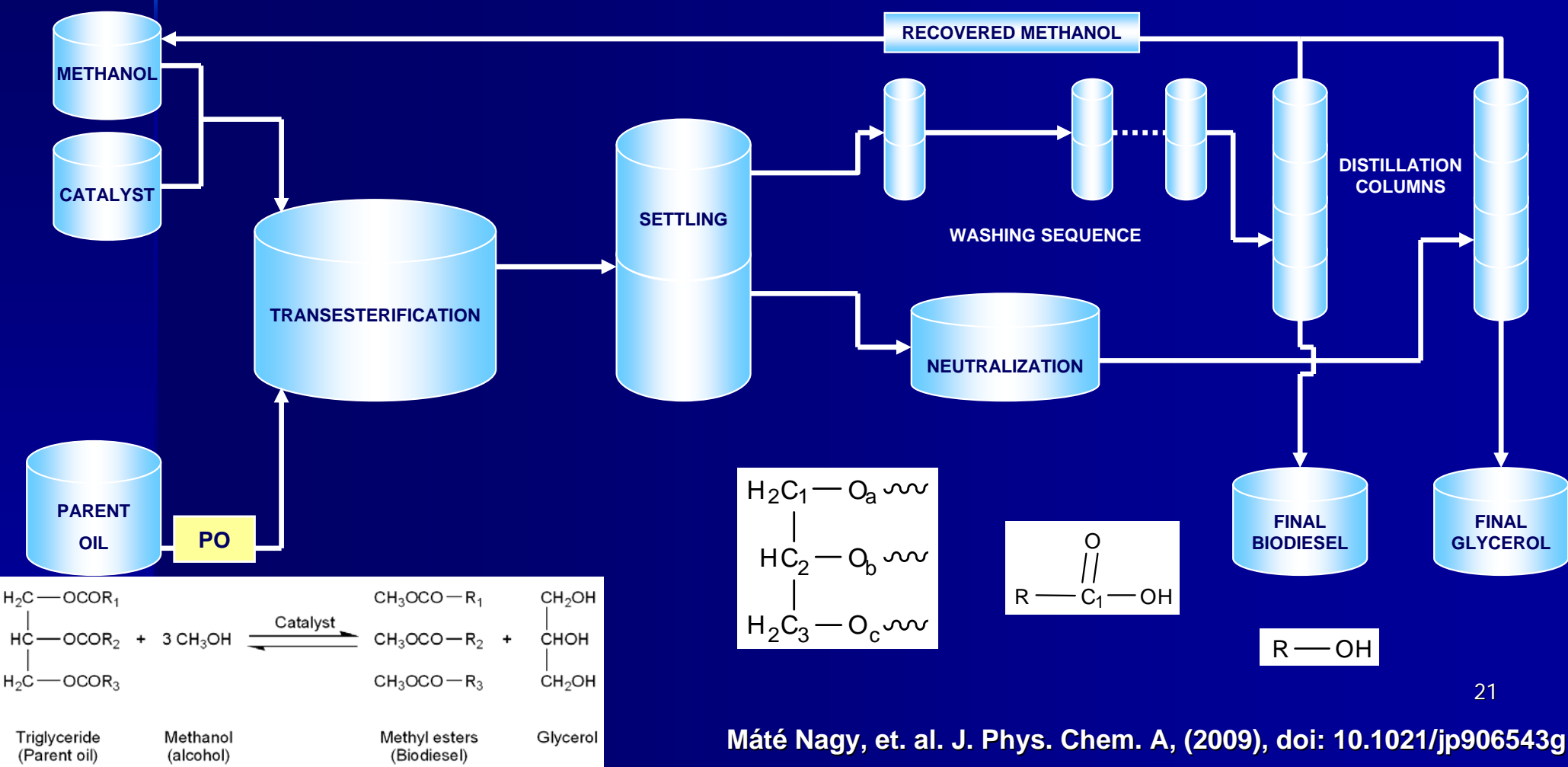
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Rapid sample analysis

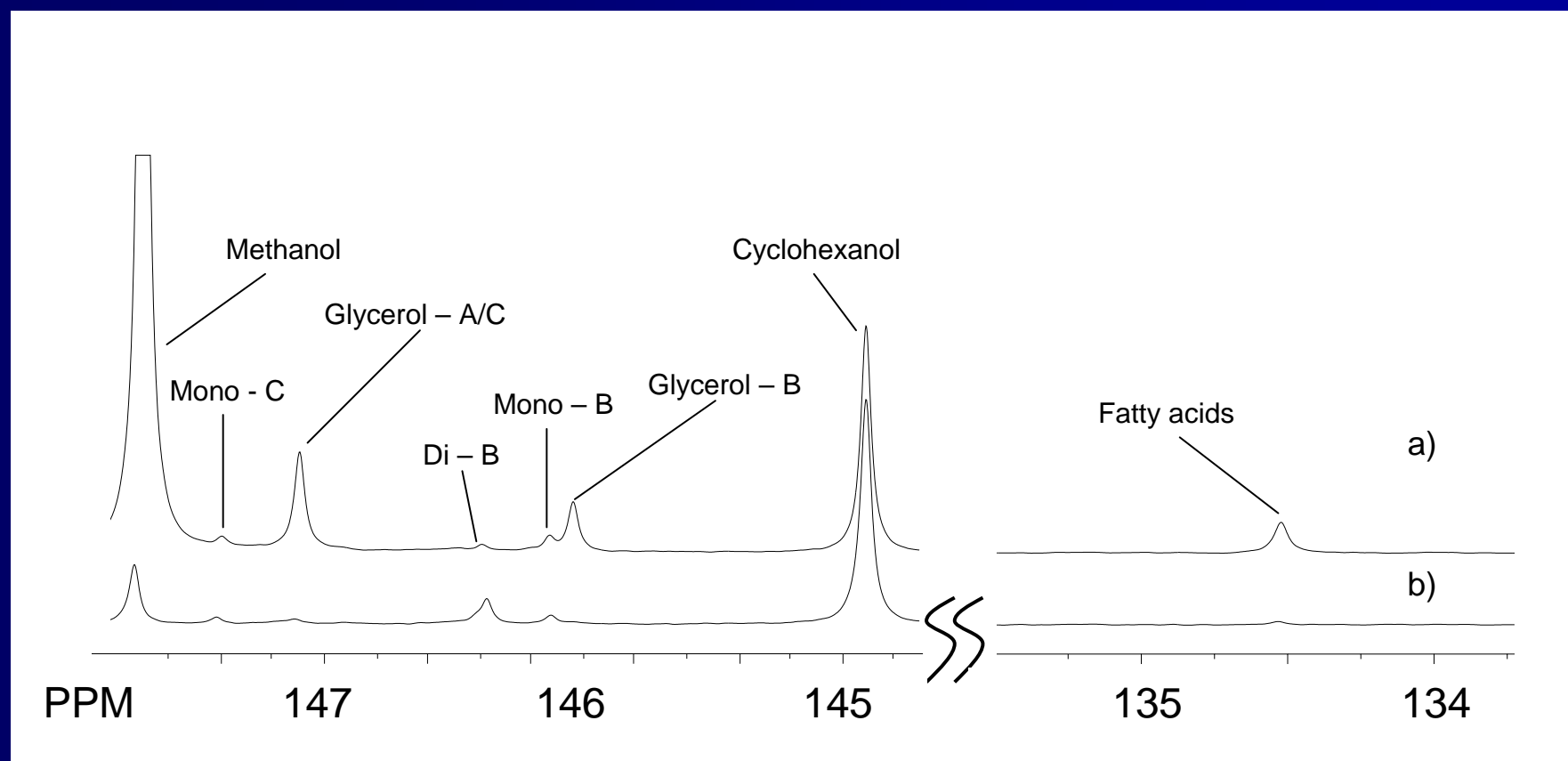
Biodiesel Analysis

Process outline and sampling



Biodiesel Analysis

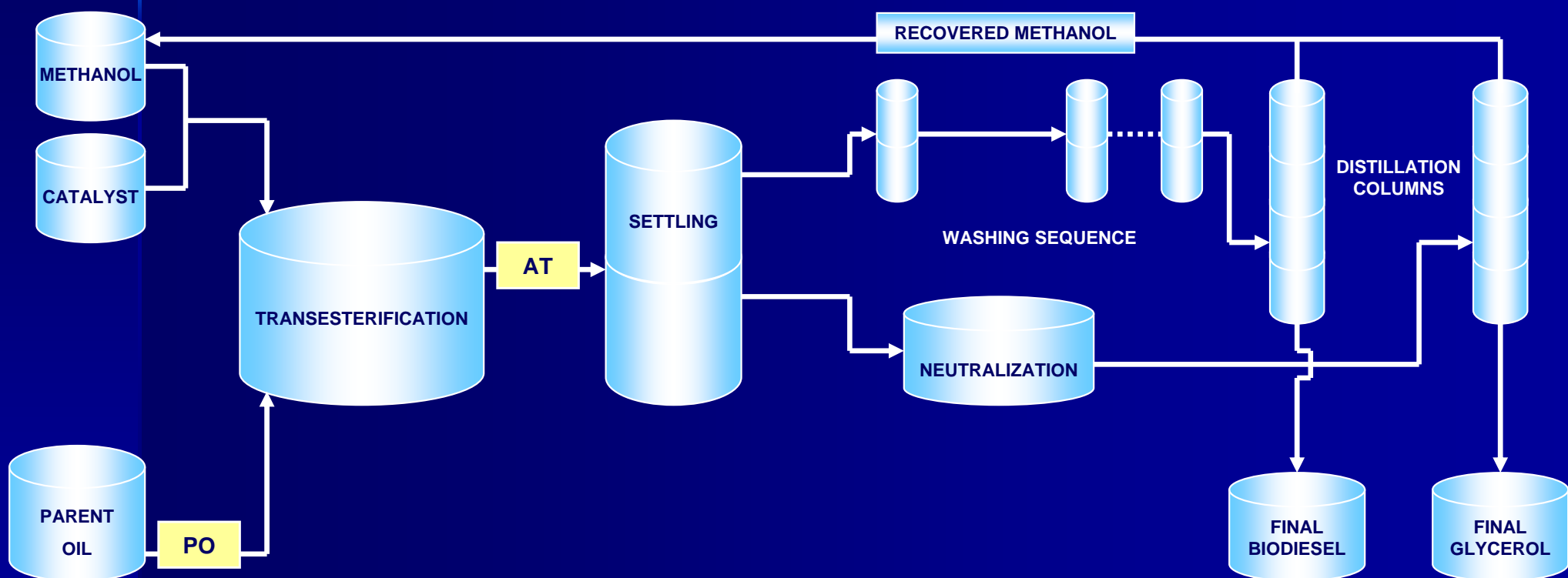
The parent oil



Quantitative ^{31}P -NMR spectra of parent oil samples of waste vegetable oil (a) and soybean oil (b)

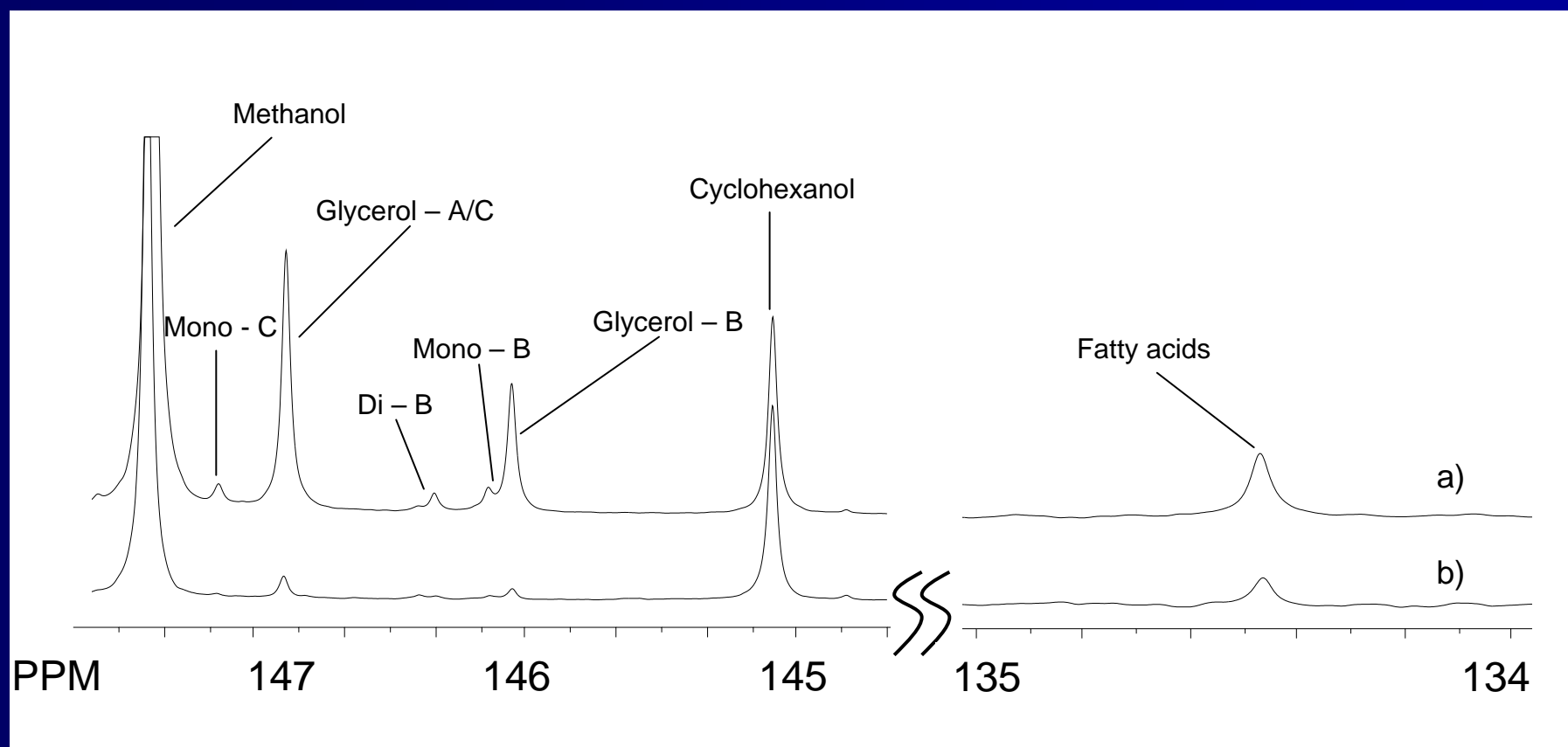
Biodiesel Analysis

Process outline and sampling



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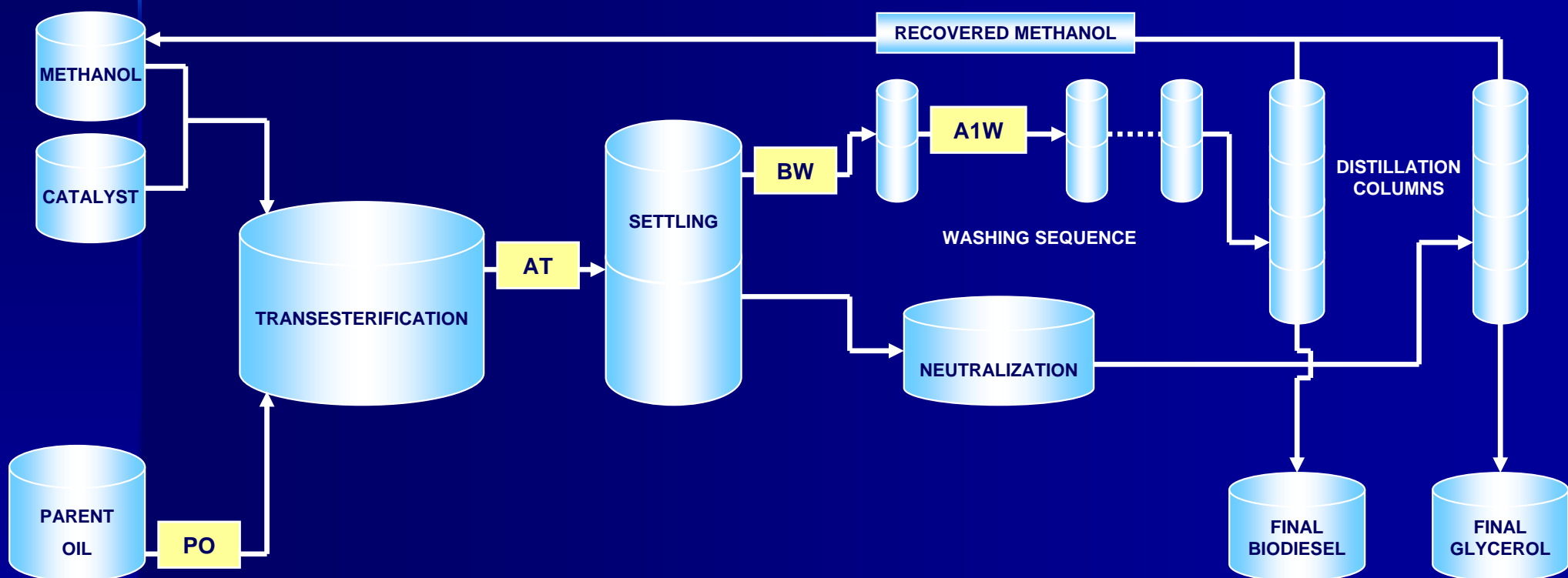
The transesterification step



Quantitative ^{31}P -NMR spectra of samples taken after the transesterification step utilizing different process conditions using soybean oil as feedstock

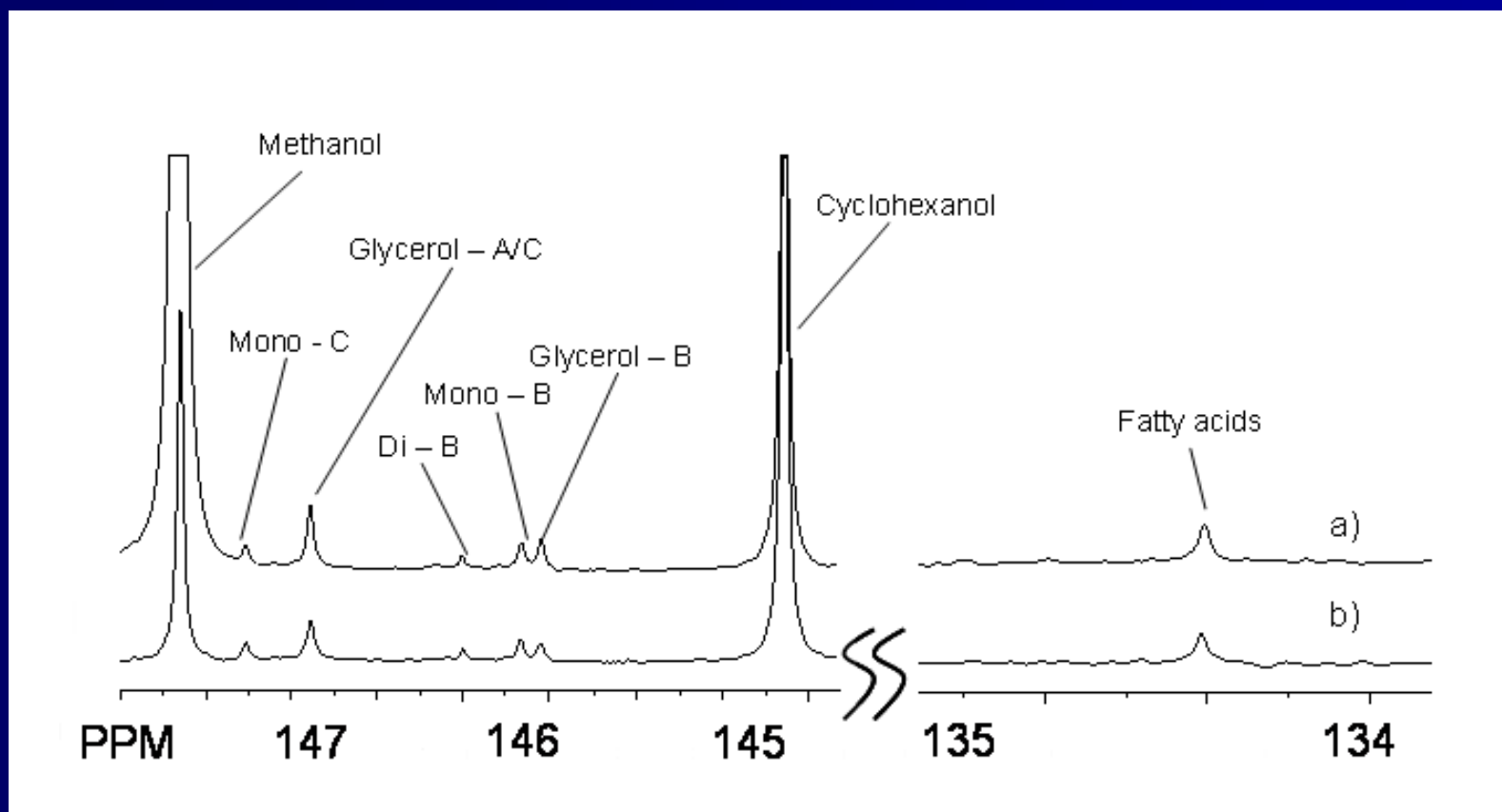
Biodiesel Analysis

Process outline and sampling



Biodiesel Analysis

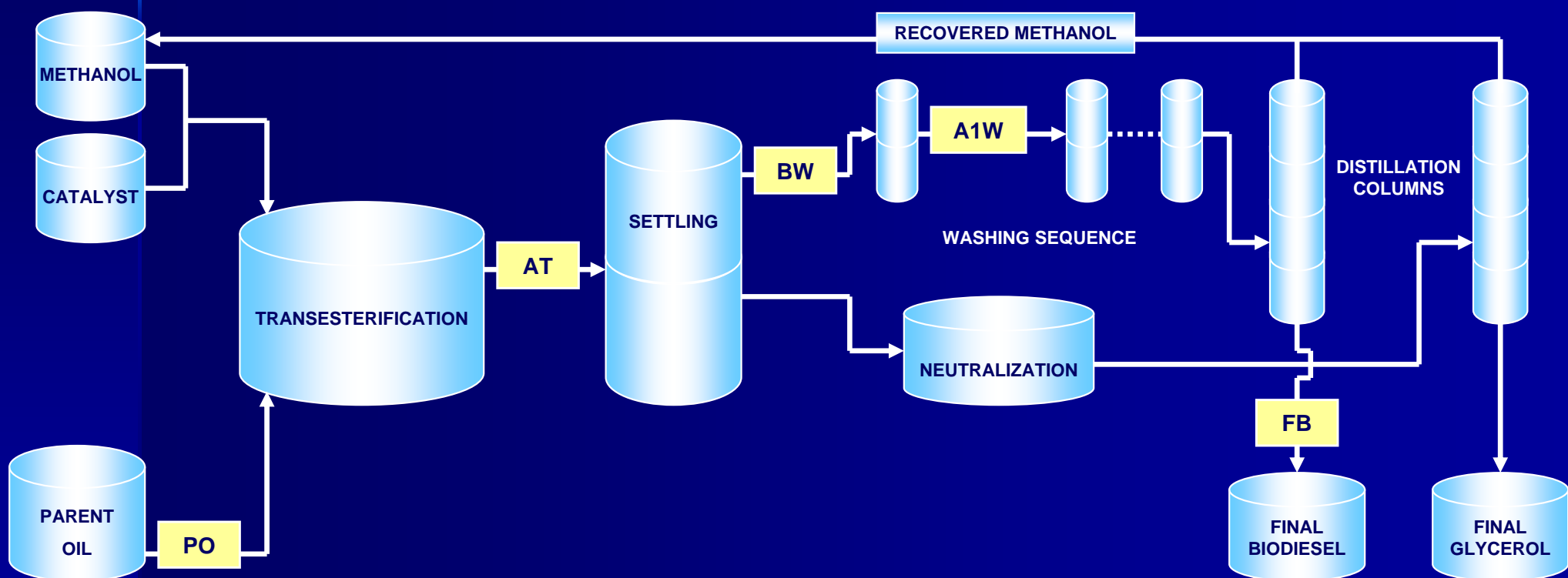
The washing phase



Quantitative ^{31}P -NMR spectra showing the washing efficiency of a soybean oil based commercial process on samples taken before wash (a) and after the first washing cycle (b)

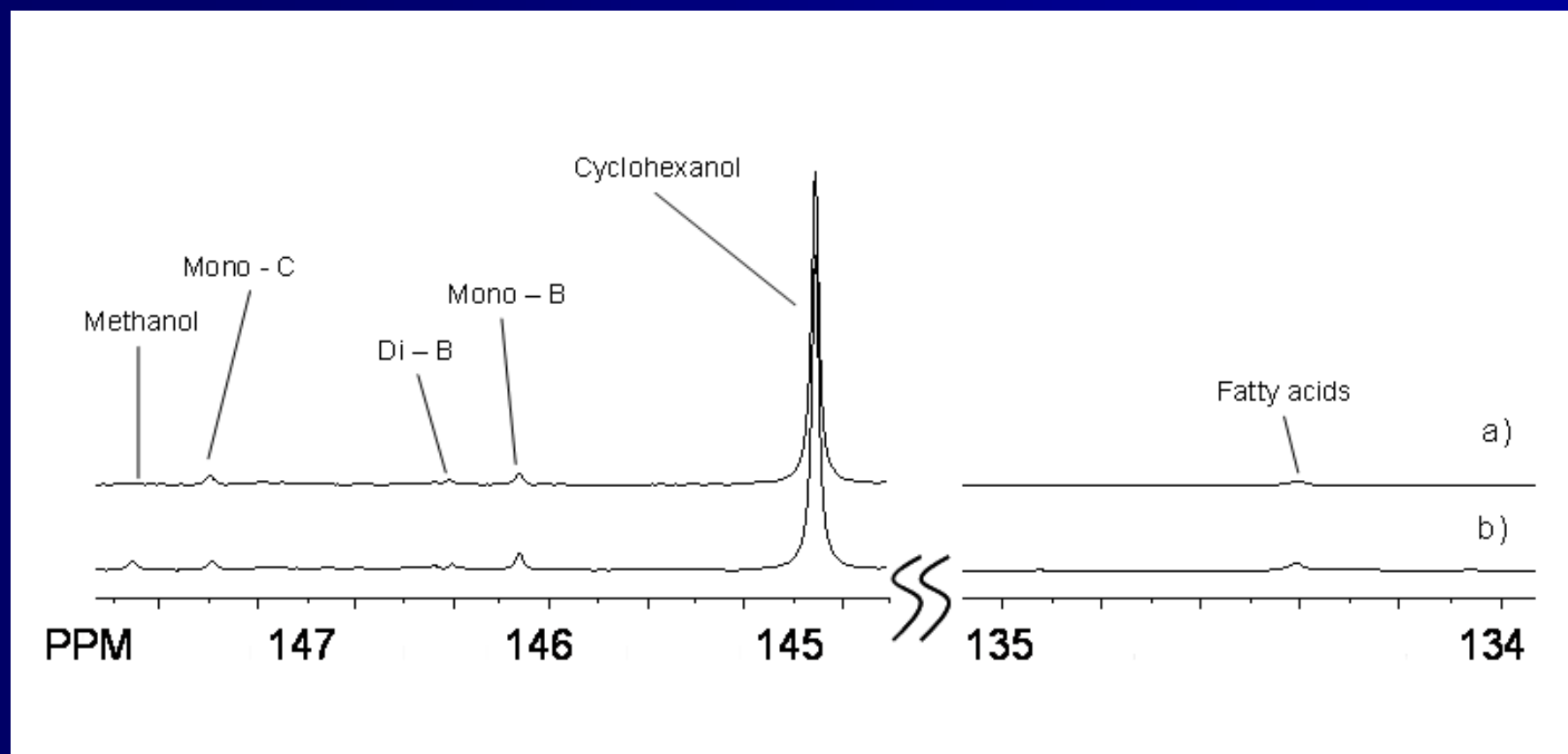
Biodiesel Analysis

Process outline and sampling



Biodiesel Analysis

The final biodiesel

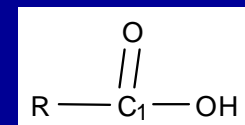
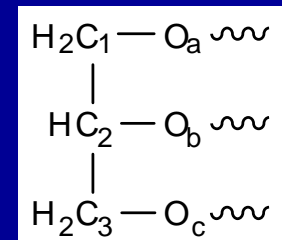
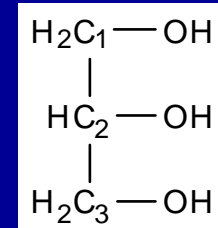
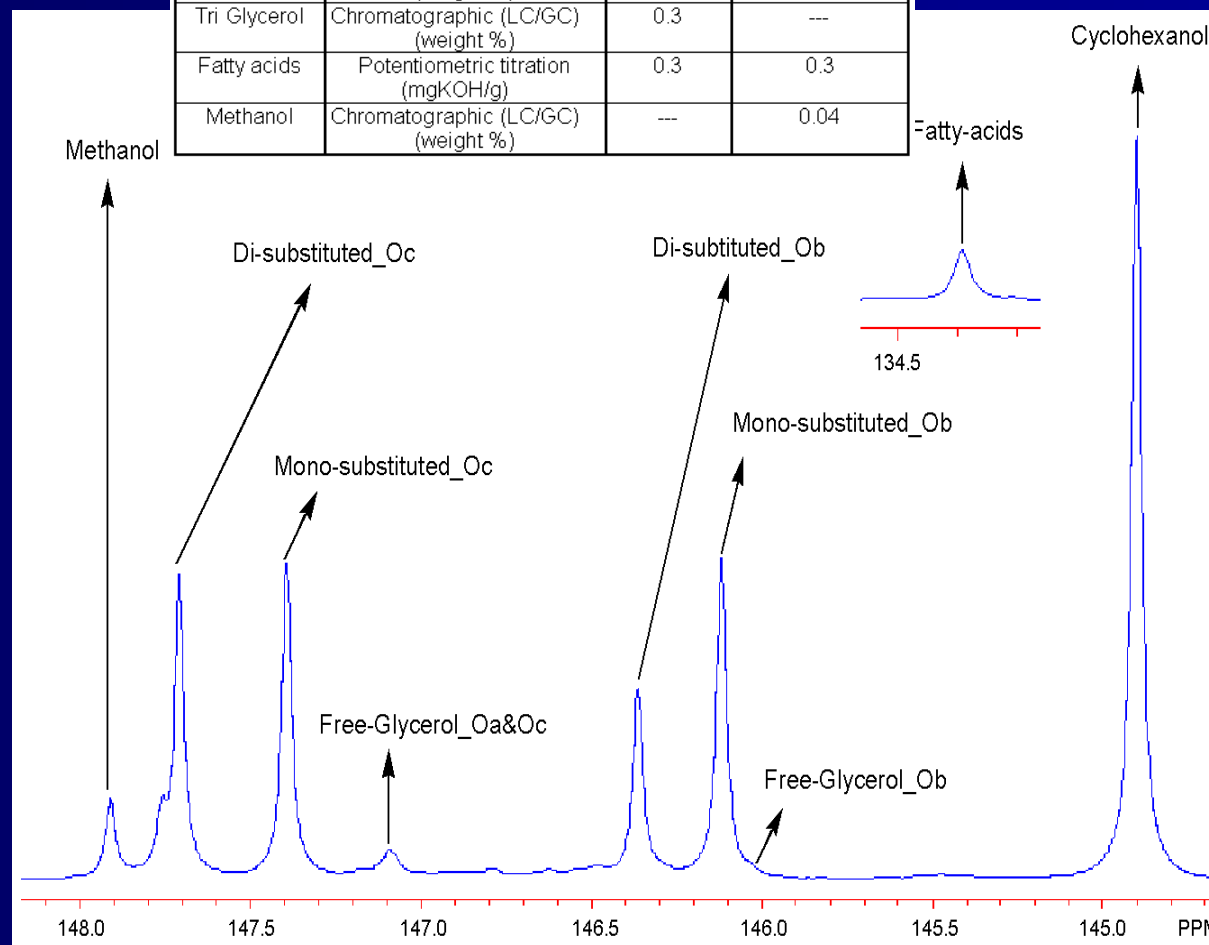


Quantitative ^{31}P -NMR spectra of final biodiesel samples of soybean oil (a) and waste vegetable oil (b)

Biodiesel Analysis

Quantitative analysis on final biodiesel sample

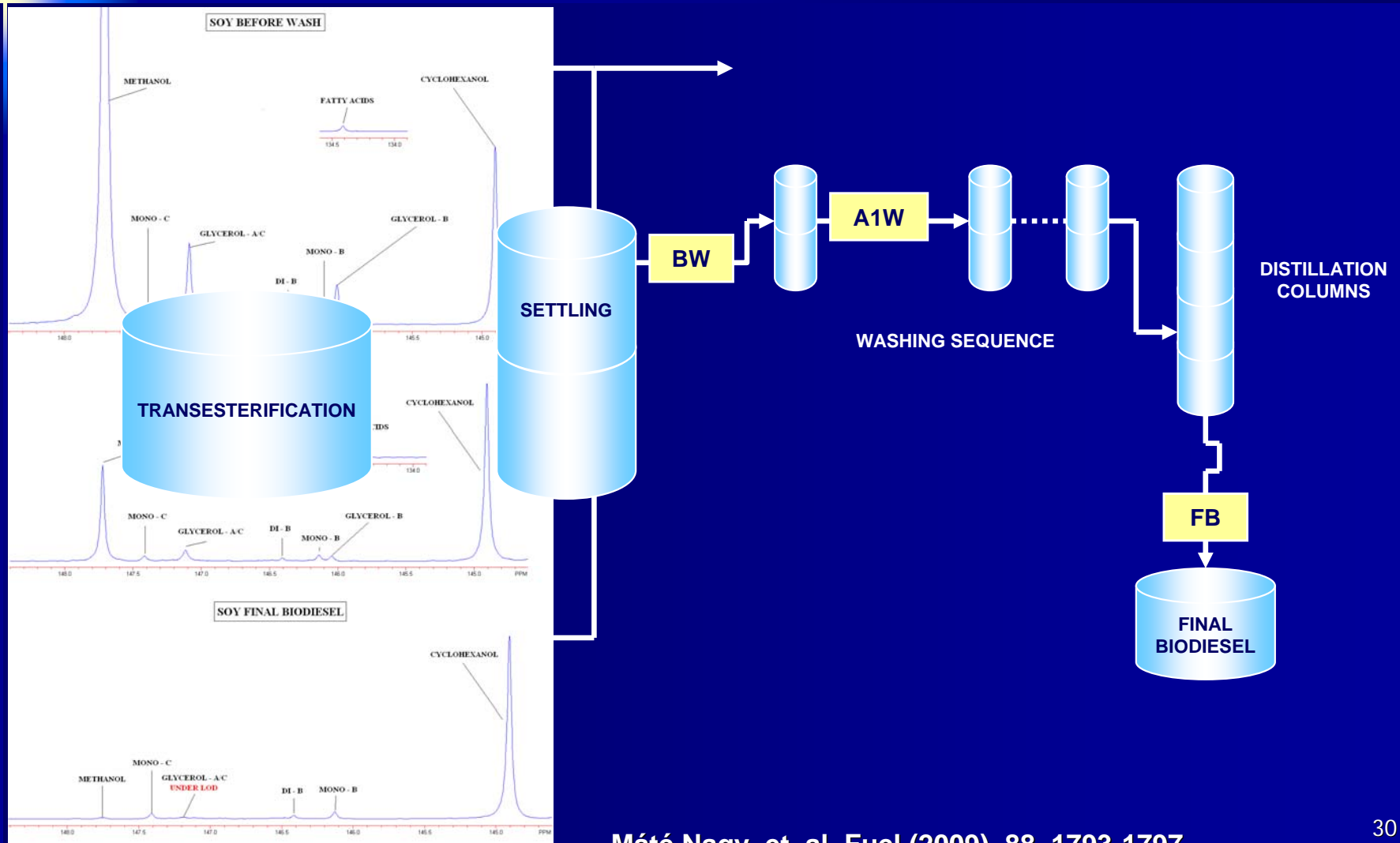
Component	Conventional technique used	Data Measured	Phosphitylation with ^{31}P -NMR
Free Glycerol	Chromatographic (LC/GC) (weight %)	0.01	0.03
Mono Glycerol	Chromatographic (LC/GC) (weight %)	1.4	1.8
Di Glycerol	Chromatographic (LC/GC) (weight %)	1.7	1.9
Tri Glycerol	Chromatographic (LC/GC) (weight %)	0.3	---
Fatty acids	Potentiometric titration (mgKOH/g)	0.3	0.3
Methanol	Chromatographic (LC/GC) (weight %)	---	0.04



Quantitative ^{31}P -NMR spectra of biodiesel samples from commercial biodiesel operations based on soy oil

Biodiesel Analysis

Monitoring the biodiesel process stream

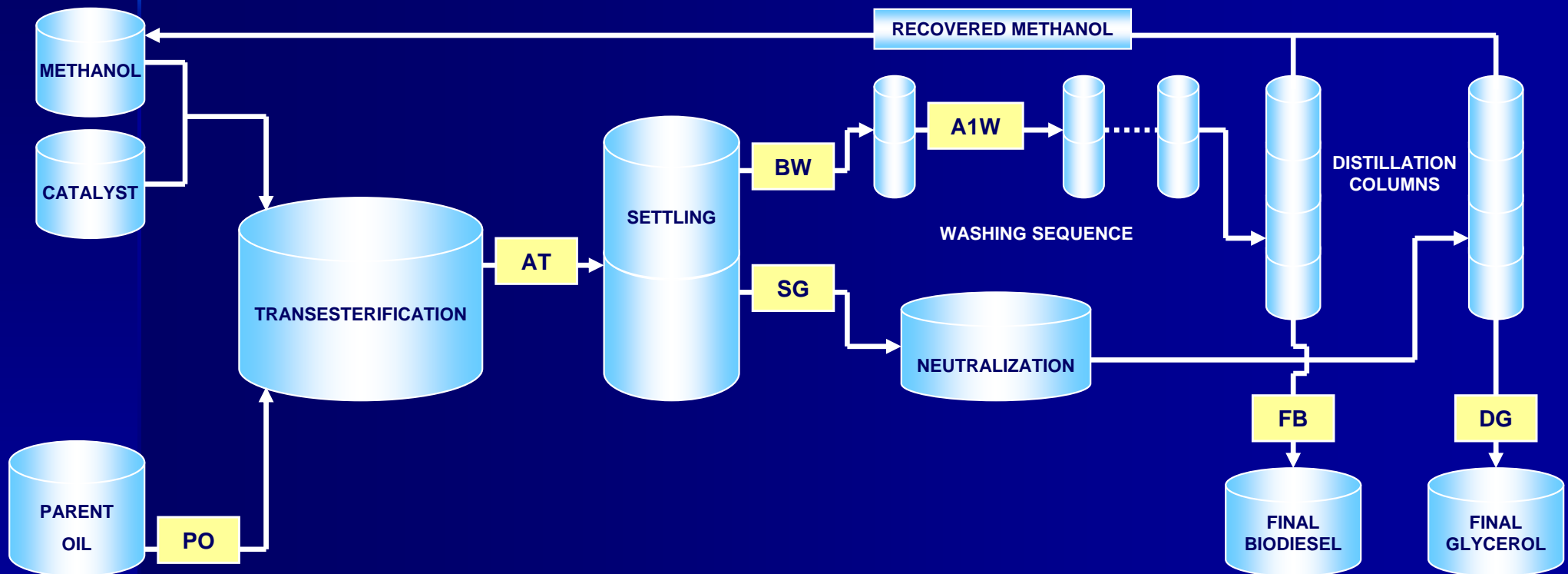


Máté Nagy, et. al. Fuel (2009), 88, 1793-1797

Máté Nagy, et. al. J. Phys. Chem. A, (2009), doi: 10.1021/jp906543g

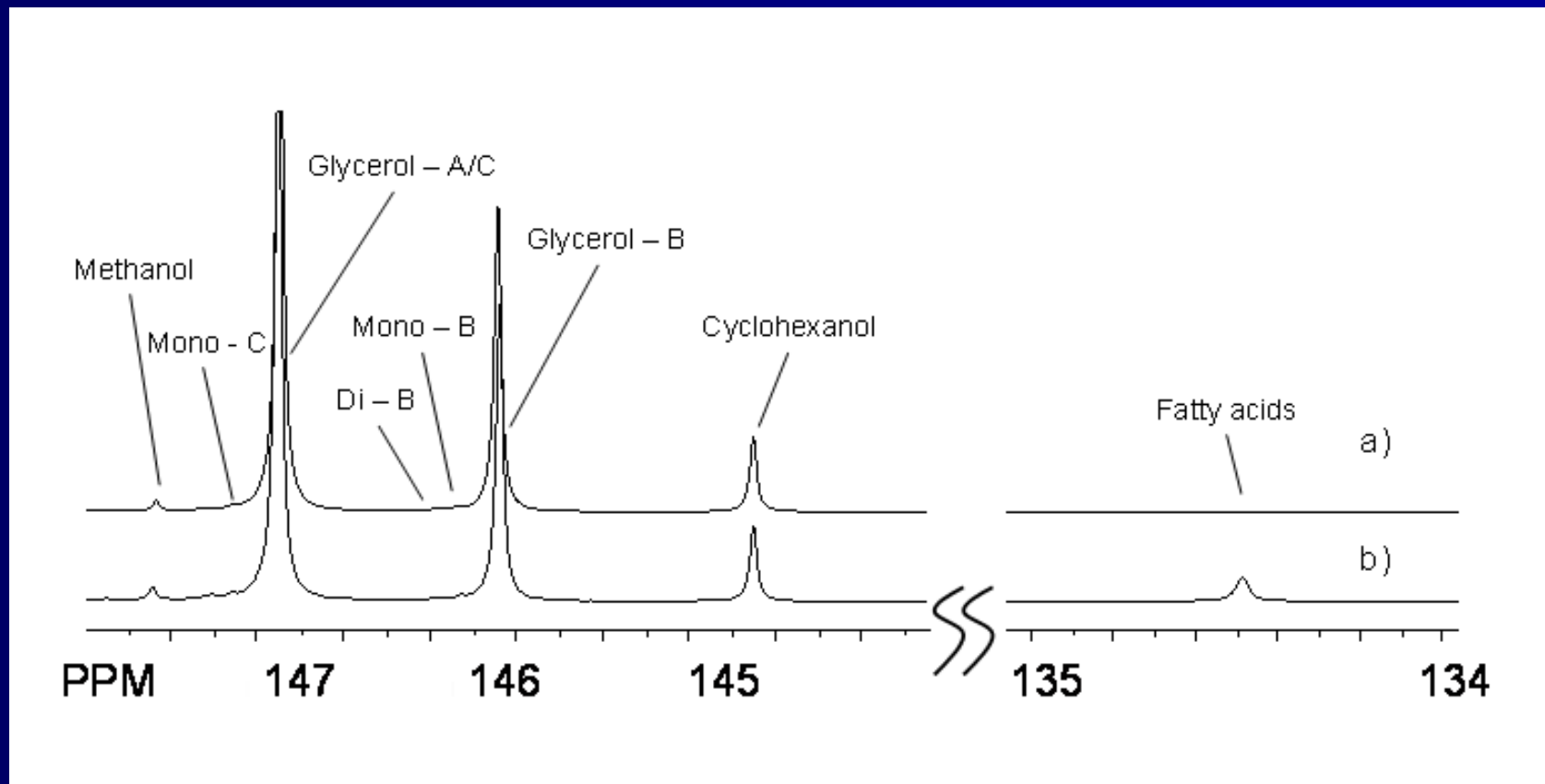
Biodiesel Analysis

Process outline and sampling



Biodiesel Analysis

The glycerol process stream

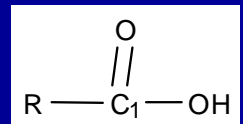
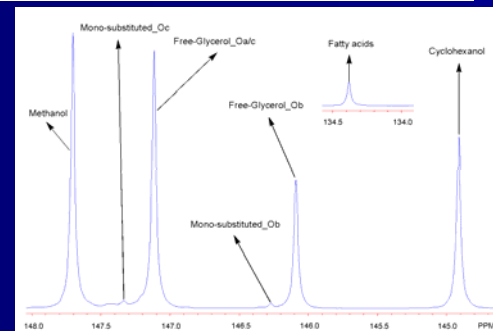
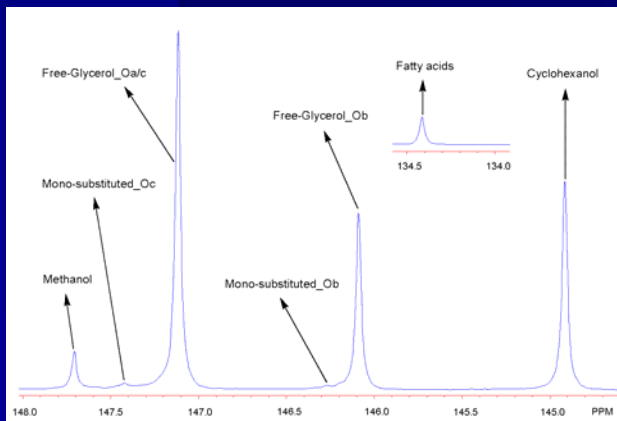
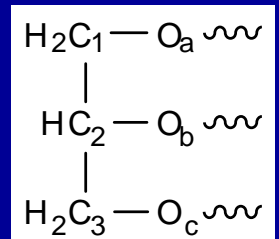
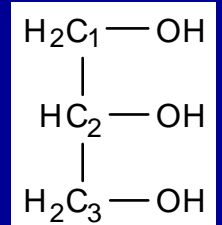
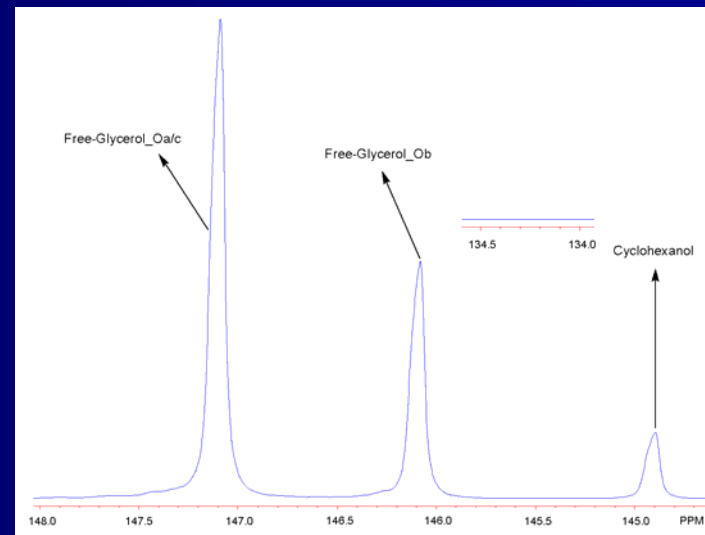
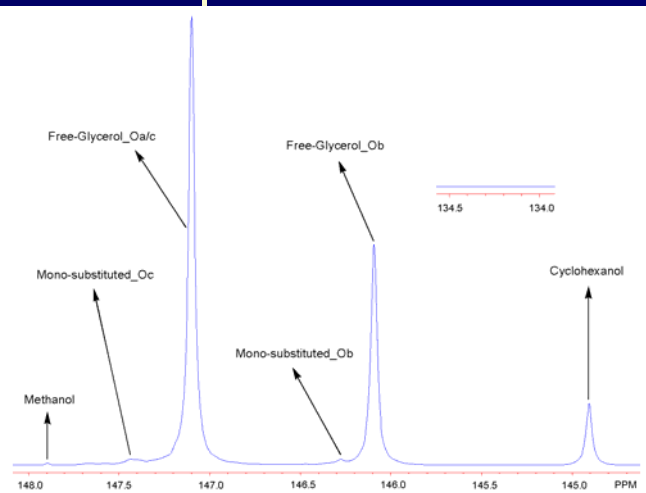


Quantitative ^{31}P -NMR spectra on glycerol samples from soybean oil based commercial process of the separated glycerol before neutralization (a) and of the final demethylated glycerol (b)

Biodiesel Analysis

Quantitative analysis on final glycerol samples

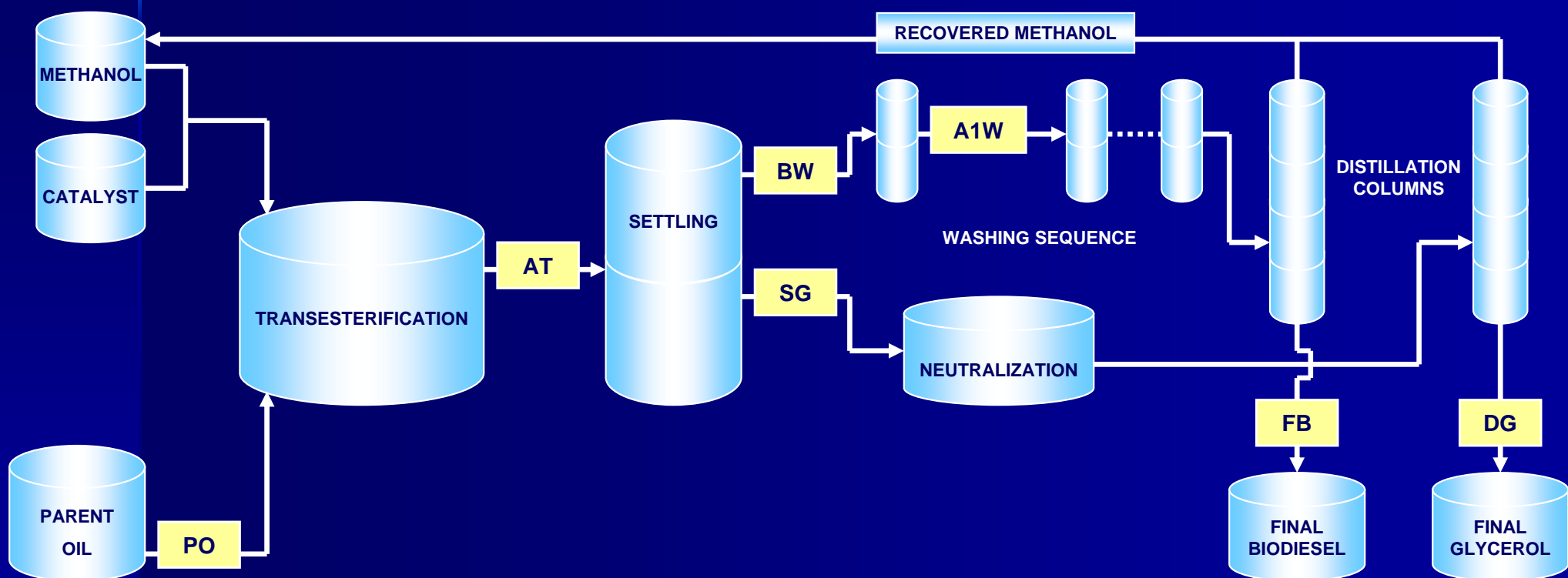
Component	Technique used	
	Conventional (weight %)	TMDP / ³¹ P-NMR (weight %)
Free glycerol	---	75.9
Mono substituted glycerol	---	7.7
Total glycerol	81.9	83.6
Fatty acids	0.0	0.0
Methanol	0.1	0.1



Quantitative ³¹P-NMR spectra of glycerol samples from commercial biodiesel operations based on soy oil, poultry fat & tallow

Biodiesel Analysis

Process outline and sampling



Conclusions

■ Current methods vs. TMDP/³¹P-NMR method

- *Fast sample preparation (under 5 min)*
- *Rapid analysis and data acquisition (56 sec)*
- *Accurate quantitative method (Accuracy: +95%)*
- *Easy data analysis (Good separation, spectroscopic information related to the structure)*
- *Sensitive novel research tool (1.9 μmol/mL LOD, ±1.1% error margin)*

■ Additional benefits

- *Independent from feedstock*
- *Direct measurement for the intermediate products through the whole production line → Process step optimization*
- *Direct measurements for contamination in the final products → Quality control*



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Thank You!

Máté Nagy

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